

The different renewable product generation via thermal processes suited to forestry raw materials

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South african context

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Wood and forestry raw materials

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Thermo-chemical processes

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Pyrolysis

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Mechanisms, **renewable products**, yields and parameters

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Toxicity

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Applications

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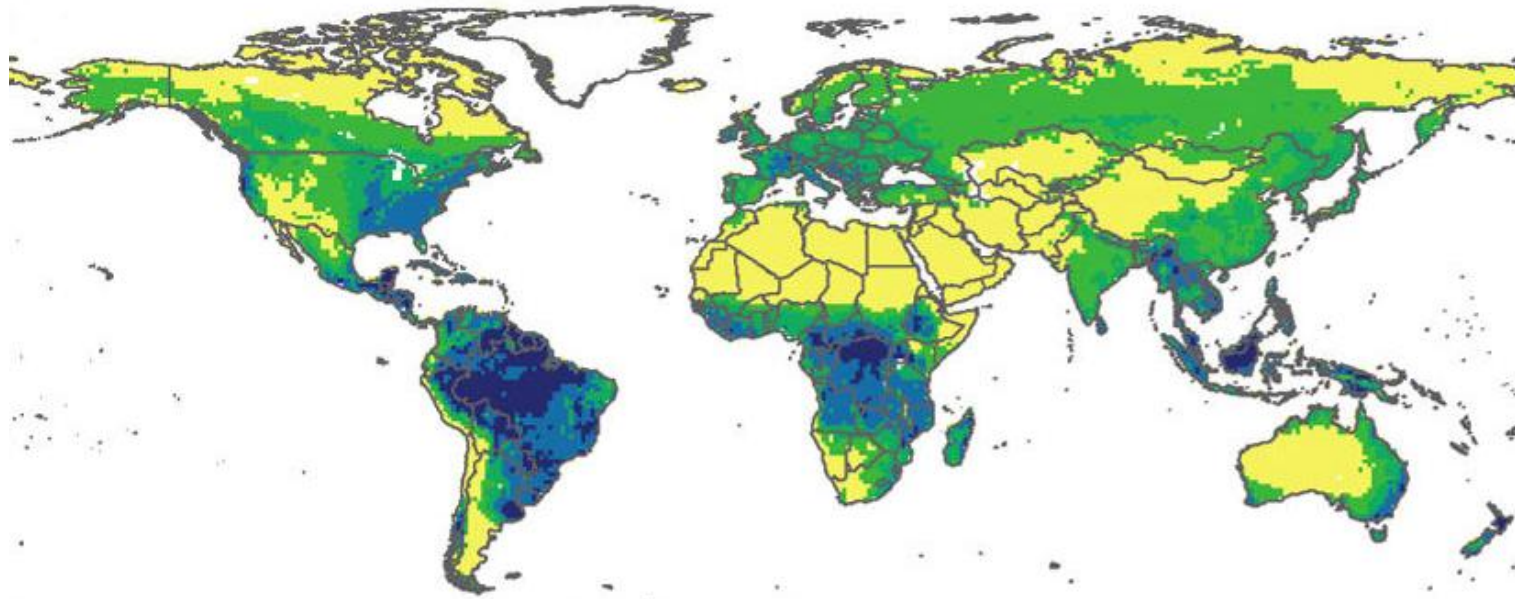
Commercial plants

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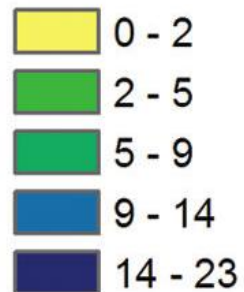
Keys-Challenges



Biomass potential



Natural Production
(ton ha⁻¹ y⁻¹)



- Large availability of natural resources
- Conversion of plant biomass is an attractive option

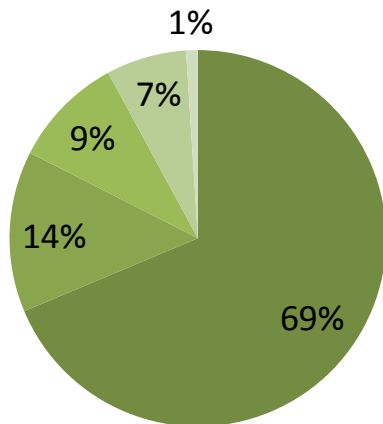
Campbell, 2008



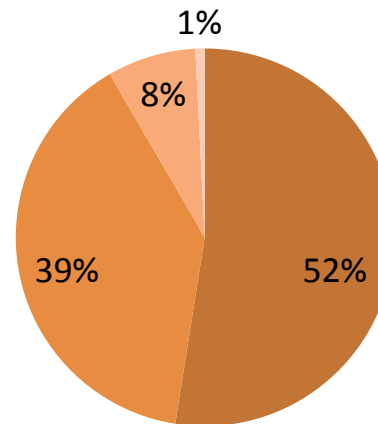
Why wood?

- One of the best renewable Carbon source (CO₂ sequestration)
- Wood resources in South Africa increasing, production of 16 m³/hectare while:
 - a decrease of plantation area from 1997
 - a small land use in South Africa

■ Grazing ■ Arable ■ Nature conservation ■ Other ■ Forestry



■ Pine ■ Eucalyptus ■ Wattle ■ Other

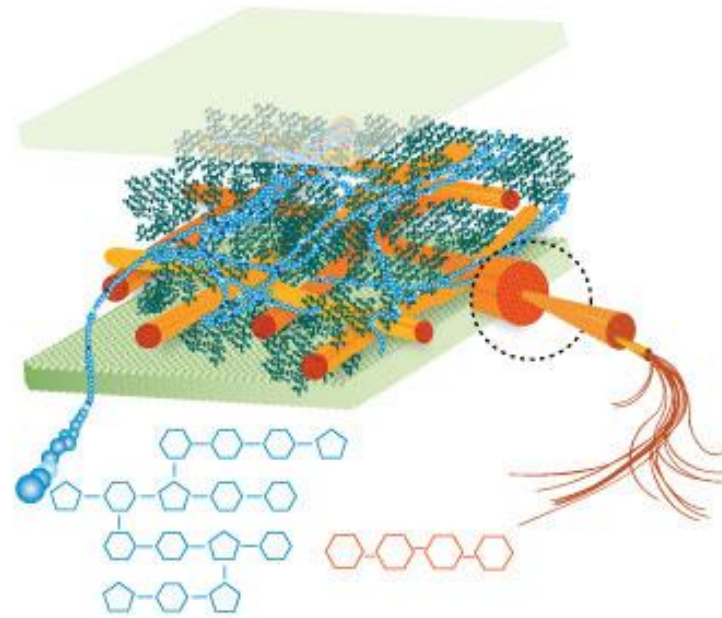


Godsmark, 2009



Why wood?

Lignocellulosic structure
which leads
to remarkable products



Hemicelluloses
Cellulose
Lignin

Woody biomass	Lignin (%)	Cellulose (%)	Hemicelluloses (%)	Reference
Pine	26.4	44.7	18.6	Hamelinck, 2005
Eucalyptus	27.7-25.9	49.5-57.3	13.1-16.8	Hamelinck, 2005; Kumar, 1992
Black Wattle	17.9-21.2	63.9	12.7	Lachke, 1988; Brown, 2007
Hardwood	16-25	40-50	35	Mohan, 2006
Softwood	23-33	40-50	28	Mohan, 2006
Pine bark	34	34	16	Arpiainen, 1989



Which type of forestry materials?

Untreated wood wastes:

- Dry urban wastes (construction and demolition waste)
- Wood from plantations Poplar, Eucalyptus, Pine
- Logging residues
- Shrubs and wood residues (twigs, branches and stumps)
- Sawdust
- Bark
- Wood chips
- Damaged wood from plantation fires



South Africa generated 13.9 tons of wood (pine, eucalyptus, wattle)/ha/year in 2008 which are mainly dedicated to: pulpwood, sawlogs, mining timber.



Why Thermo-chemical processes ?

- Carbon-neutral balance
- Lower emissions of NO_x, CO and hydrocarbons
- All the products from these processes can be used
- Improvement of efficiency of the energy conversion
- Production of sustainable energy sources
- Material cheap and readily available
- Relatively low capital investment
- Large range of potential products/feedstocks

Choice of conversion process depends upon the **type** and **quantity** of **biomass feedstock**, the **desired** form of the **energy**, end **use** requirements, environmental standards, economic conditions and project specific factors.



But ?

- **High moisture content** inhibits pyrolysis, leads to energy loss because used to evaporate water,
- Often **low density**, which makes them bulky and hard to store,
- Often **no homogeneous** form,
- Contain a fairly large amount of **inorganic substances**, which remains as **ash**,
- **Environmental impact of pre-treatments** used to separate or remove products.

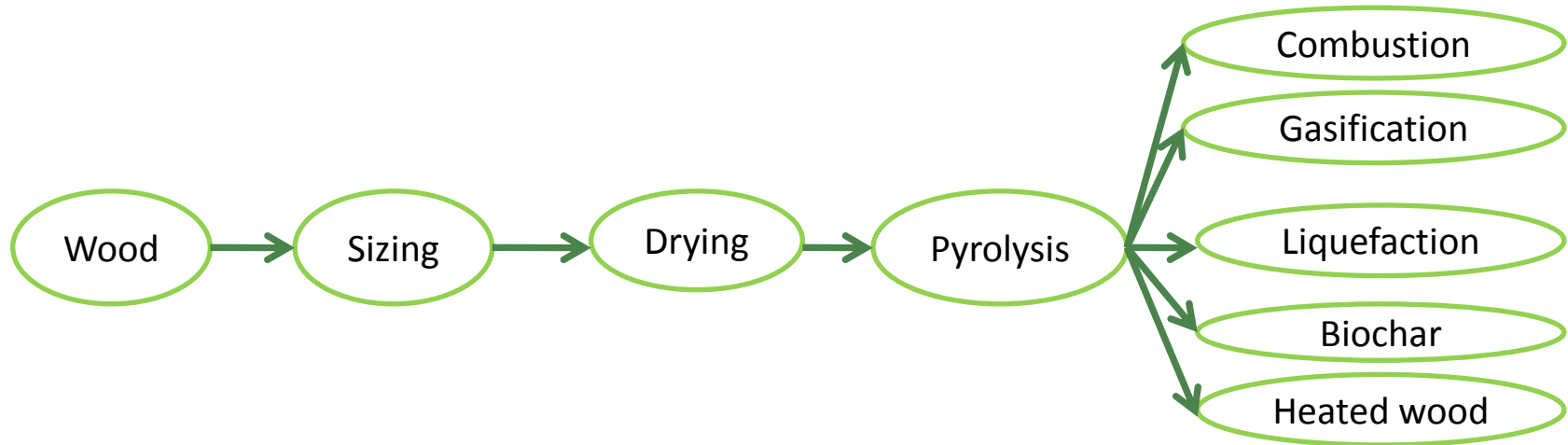


Thermo-chemical processes

Thermal Process	Temperature (°C)	Atmosphere	Products	Energy efficiency (%)
Combustion	> 900	O ₂ (air)	CO ₂ + H ₂ O + N ₂ + ashes to be treated	65
Pyrolysis	< 600	Inert gas or low pressure	Char + tars + gas, which proportions are related to the pyrolysis parameters	45
Gasification by Fast Pyrolysis	> 700	Inert gas or low pressure	Mainly gas (CO, H ₂ , CH ₄ , C ₂ H ₄ ...) with low quantity of char used	75
Gasification	> 800	Air or H ₂ O vapour	Gas (H ₂ , CO, CO ₂ , CH ₄ , N ₂)	50-60
Liquefaction by Fast Pyrolysis	< 550	Low pressure	High viscosity liquid (phenols)	75
Direct liquefaction	300-350	CO high pressure	High viscosity liquid (phenols) non soluble in water	80



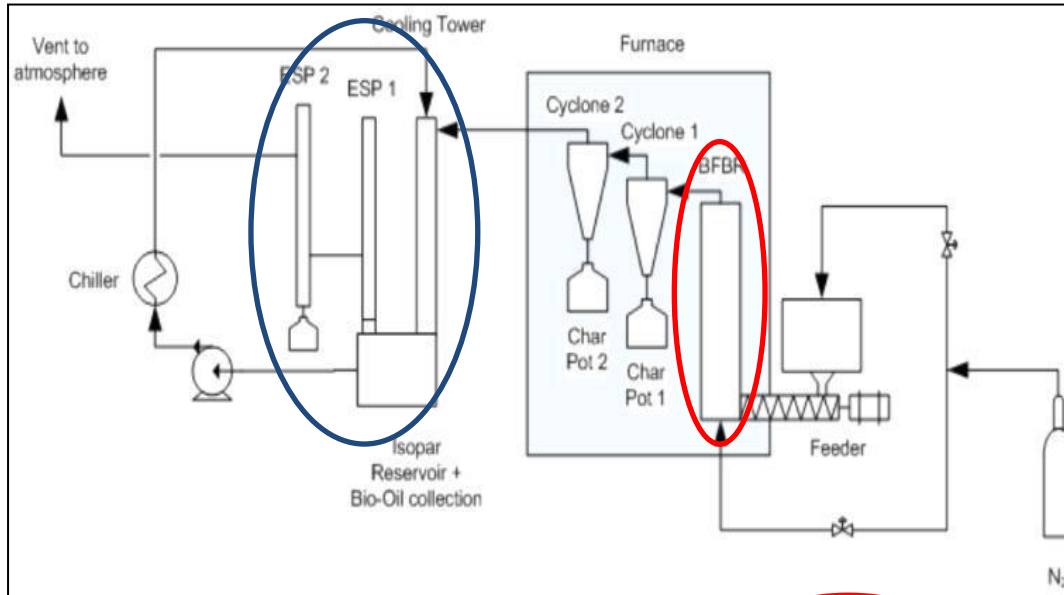
Pyrolysis, the key



Process	Temperature (°C)	Pressure (kPa)	Heating rate (°C min ⁻¹)	Residence time	Particle size	Atmosphere
Fast	450-600	101	100-300	2 s	2-5 mm	N ₂ or Ar
Slow	400-500	101	5-50	5-30 min	Large	N ₂ or Ar
Vacuum	350-500	10	5-50	2-30 s	large	Vacuum



Pyrolytic reactors



Fast pyrolysis

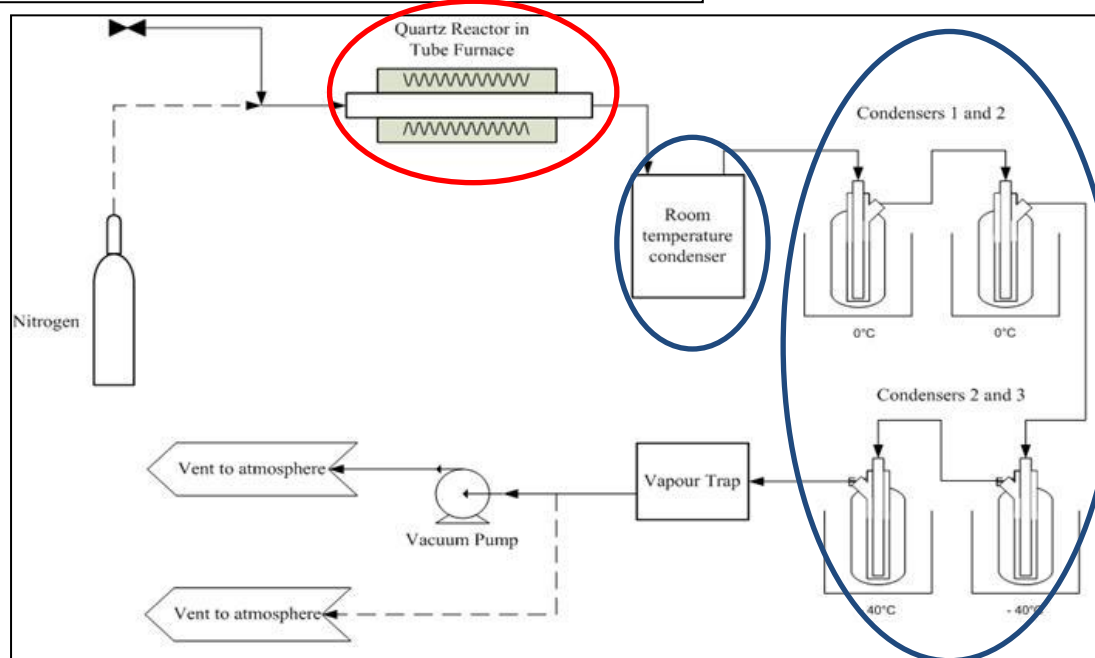
Bubbling Fluidized Bed reactor

Electrostatic separator

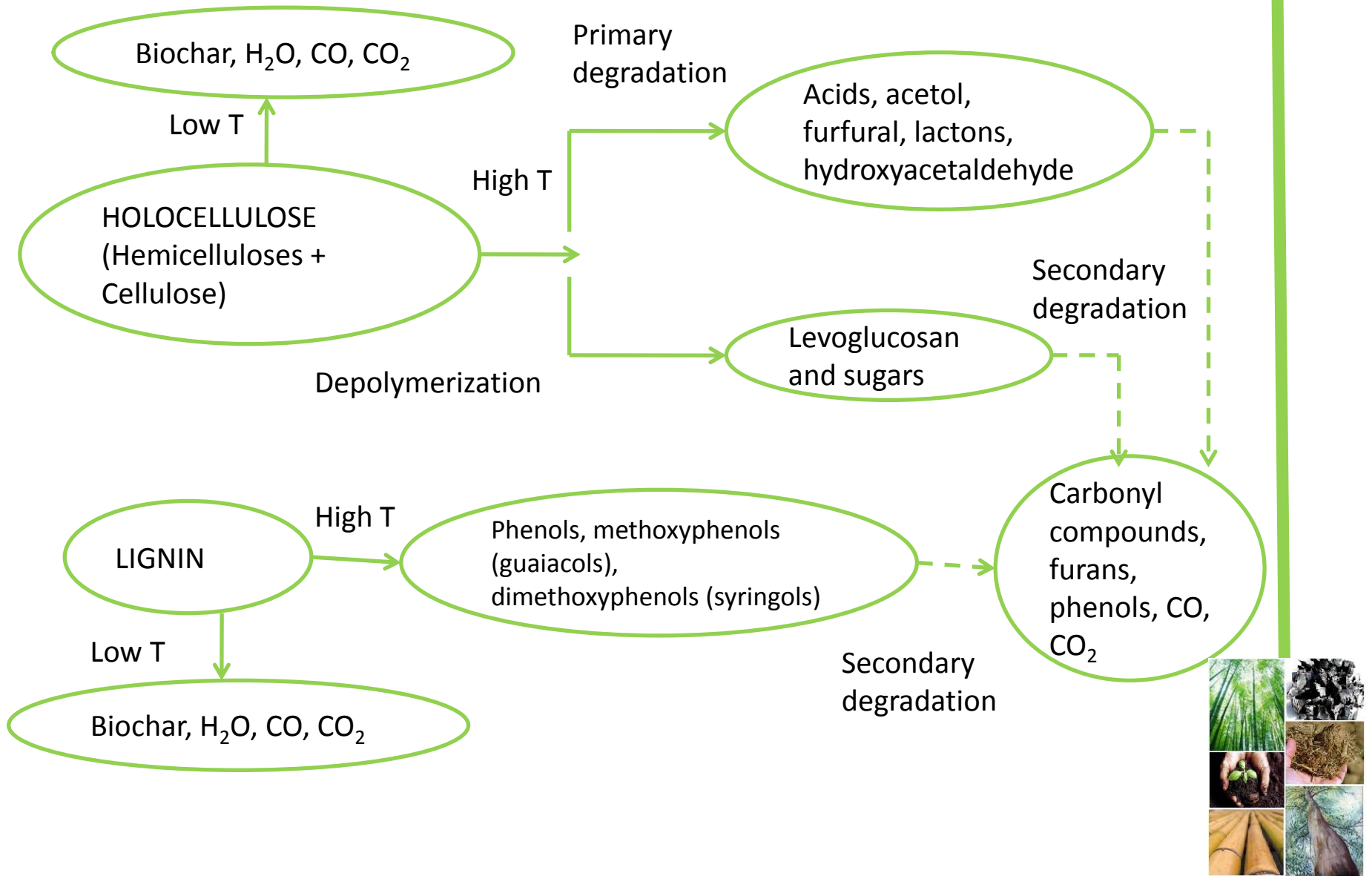
Slow/Vacuum pyrolysis

Fixed bed reactor

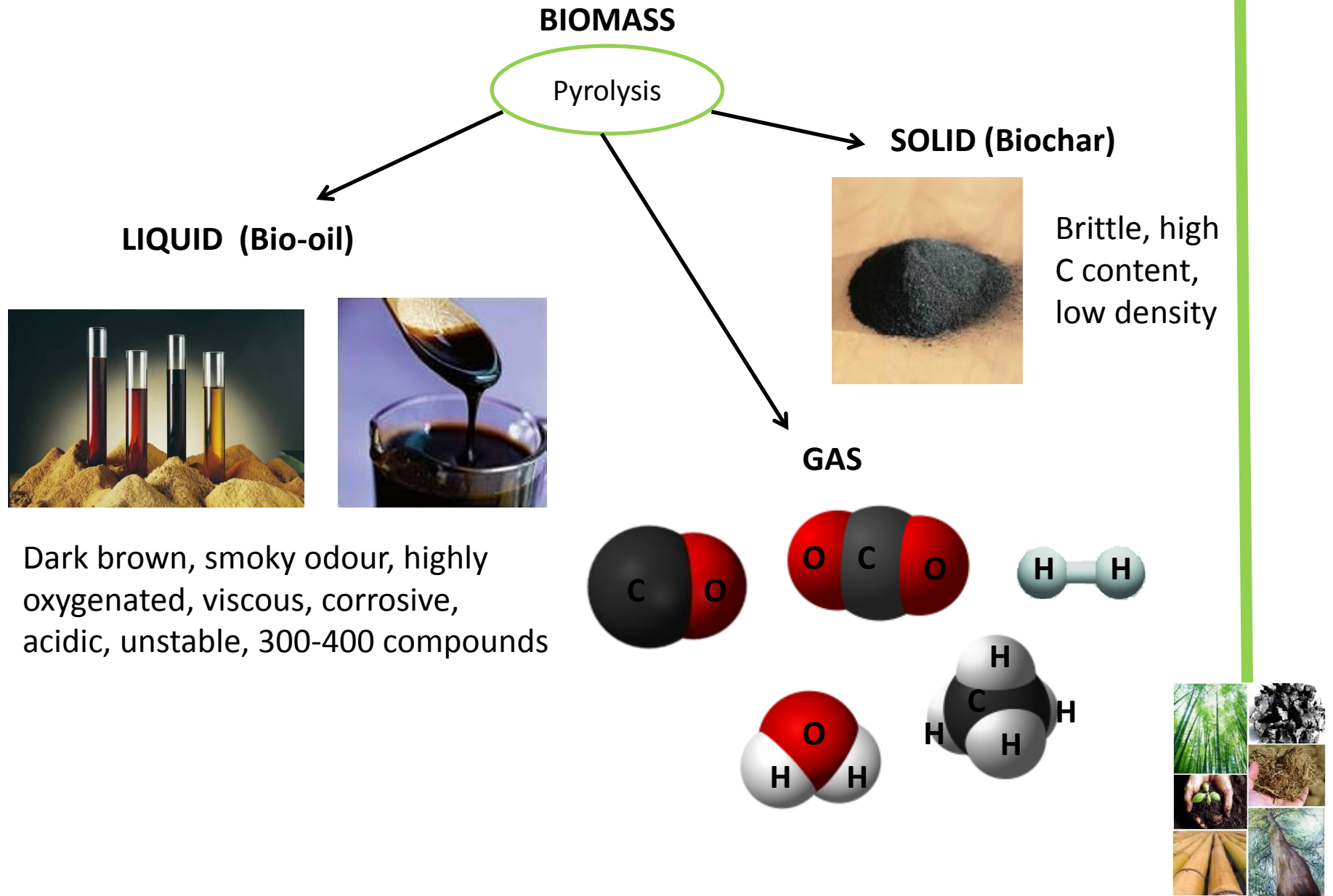
Series of condensers



Mechanisms of degradation



Renewable products



Yields of renewable forestry products

Wood biomass	Process	YLiquid (wt.%)	YBiochar (wt.%)	YGas (wt.%)	Reference
100 % white wood	FP	70	18	12	<i>Dynamotive, 2001</i>
75/25 white wood/bark	FP	65	22	13	<i>Dynamotive, 2001</i>
50/50 white wood/bark	FP	63	23	14	<i>Dynamotive, 2001</i>
100 % bark	FP	58	27	15	<i>Dynamotive, 2001</i>
Eucalyptus	FP	76	13	11	<i>Dynamotive, 2001</i>
Eucalyptus	FP	71	29		<i>Oasma, 2010</i>
Eucalyptus	SP	46	38	16.0	<i>Pimenta, 1998</i>
Pine	FP	65-75	5-18	5-20	<i>Wagenaar, 1993</i>
Pine	SP	21-30	23-36	11-23	<i>Sensoz, 2002</i>
Black wattle	VP	27	38	35	<i>Uras, 2010</i>
Softwood bark	VP	23-28	57-57	15-21	<i>Darmstadt, 2000</i>

- Maximization of bio-oil yield for Fast Pyrolysis
- Influence of the feedstock nature
- Good distribution in product yields for Slow and Vacuum pyrolysis



Influence of parameters

Size of particles (SP)
Moisture content (MC)
Ash content (AC)

Temperature (T)
Pressure (P)
Heating rate (HR)
Residence time (RT)
Flow of nitrogen (FN)

	SP (μm)	MC	AC	T ($^{\circ}\text{C}$)	P (kPa)	HR ($^{\circ}\text{C min}^{-1}$)	RT	FN
Ybio-oil	< 450 (-)	Slightly \nearrow cont. (+)	High cont. (-)	500-900 (-)	(+)	Higher rate (+)	Long time (=)	(=/+)
HHV oil		Slightly \nearrow cont. (+)			10 (-)	Low rate (+)	Long time (=)	
Ychar	2500 (+)	Slightly influence	High cont. (-)	500-900 (-)	(+)	Low rate (+)	Long time (-)	High flow (-)
HHV char			High cont. (-)	500-700 (+)	(-)	Low rate (+)	Long time (+)	High flow (-)
BET surface area			High cont. (-)	350-500 (+)	(-)	Optimal rate ~10	Optimal time~2h	

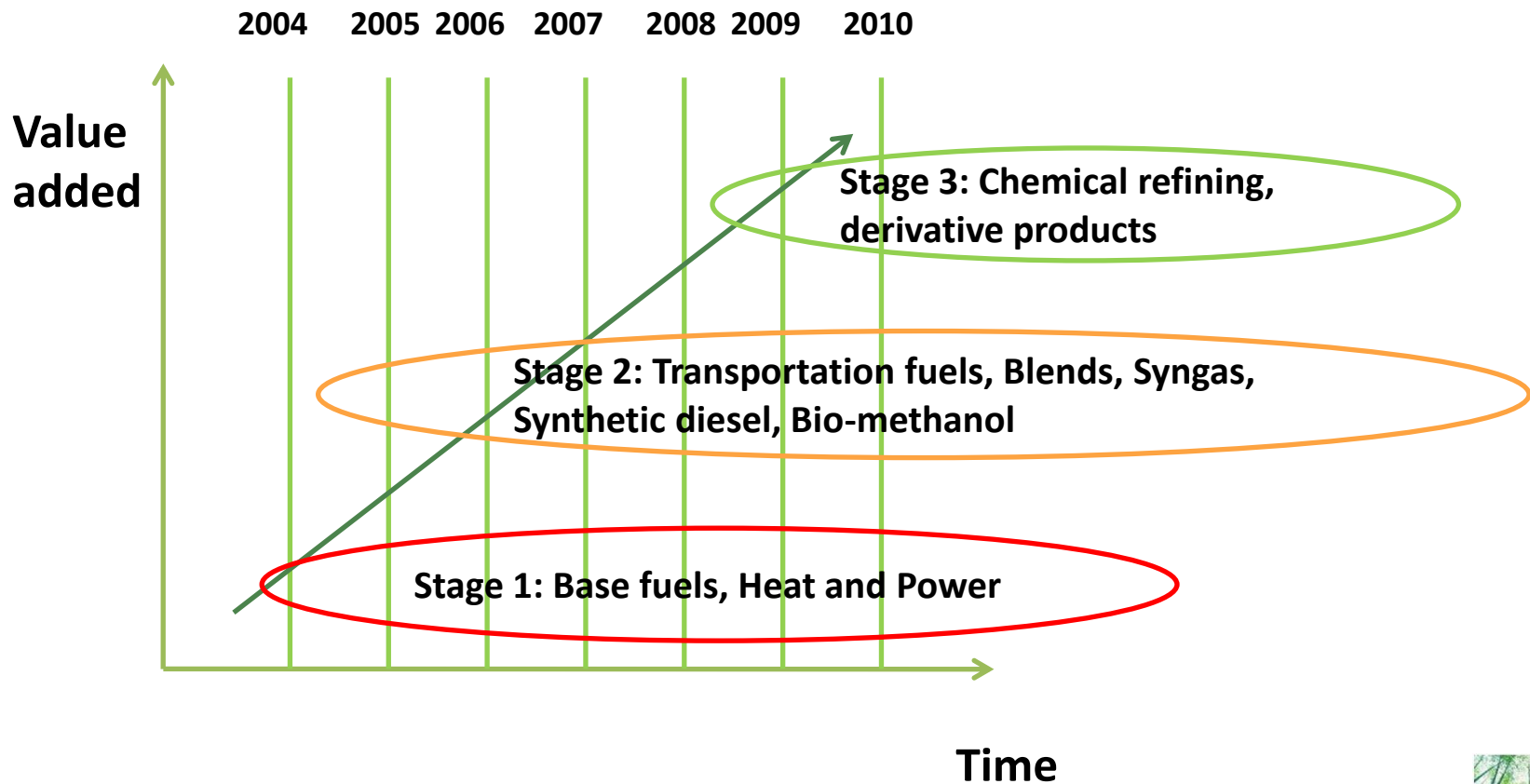


Toxicity of renewable products

Product	Health-Environmental Impacts
Bio-oil	<ul style="list-style-type: none">- Presence of Polycyclic Aromatic Hydrocarbons (PAH)- Mutagenecity tests (Salmonella strains) with slow and fast bio-oils: Slow toxic- Eutrophication test: Low fertilising effect.- Ecotoxicologic effect: Acute toxicity of slow pyrolysis oil on Daphnia- Corrosive, moderate sensitizer
Biochar	Presence of heavy metals which are toxic
Gases	
CO ₂ , CxHy	- Green House Gases contribution
CO, H ₂ S, NOx	- Toxic gas emission
HCl, HF, SO ₂	- Acidic rain
Ash	- Serious health hazard, because the nano-particles are easily inhaled and can cause lung damage.



Applications of thermo-chemical processes



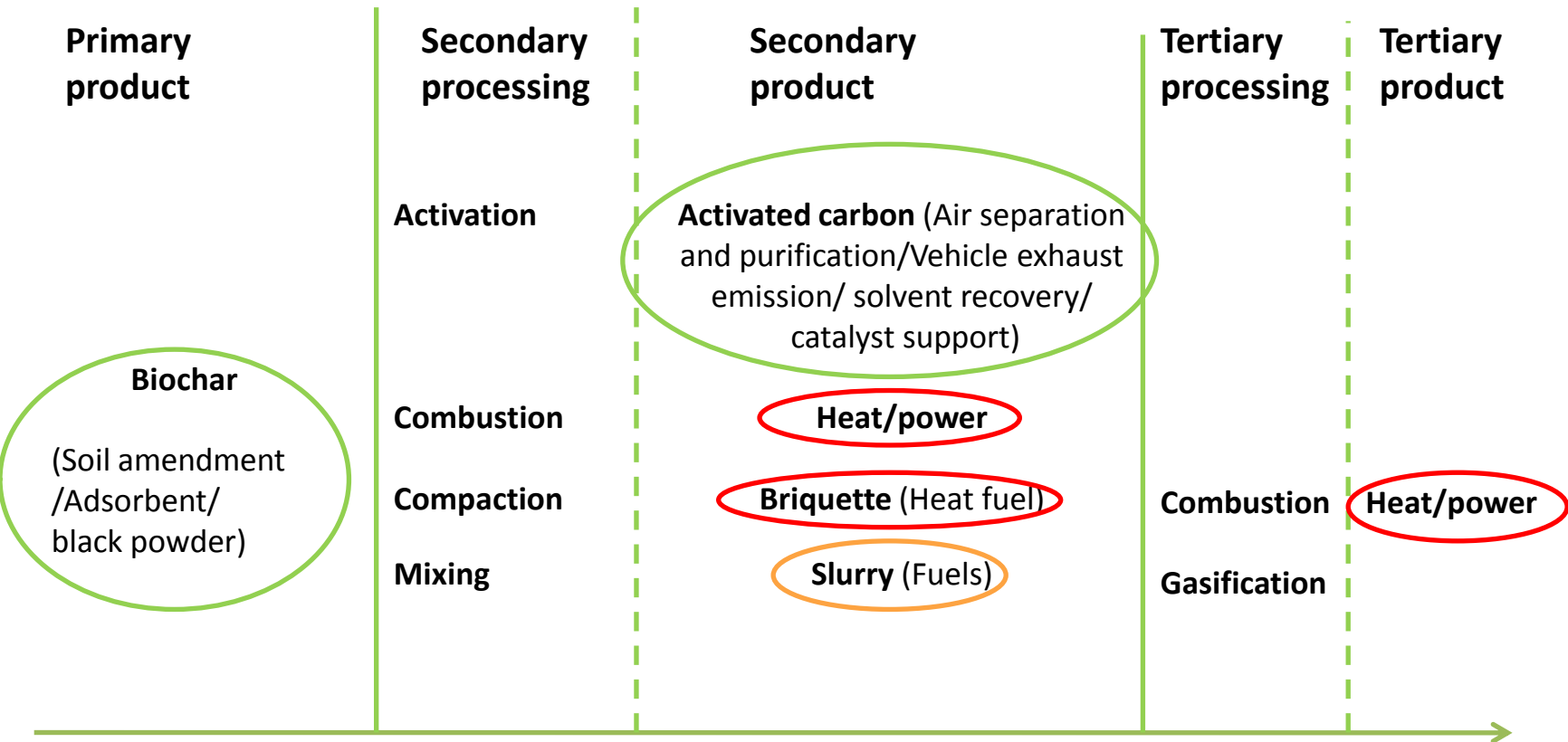
Stage 1

Stage 2

Stage 3



Applications of thermo-chemical processes



Stage 1

Stage 2

Stage 3

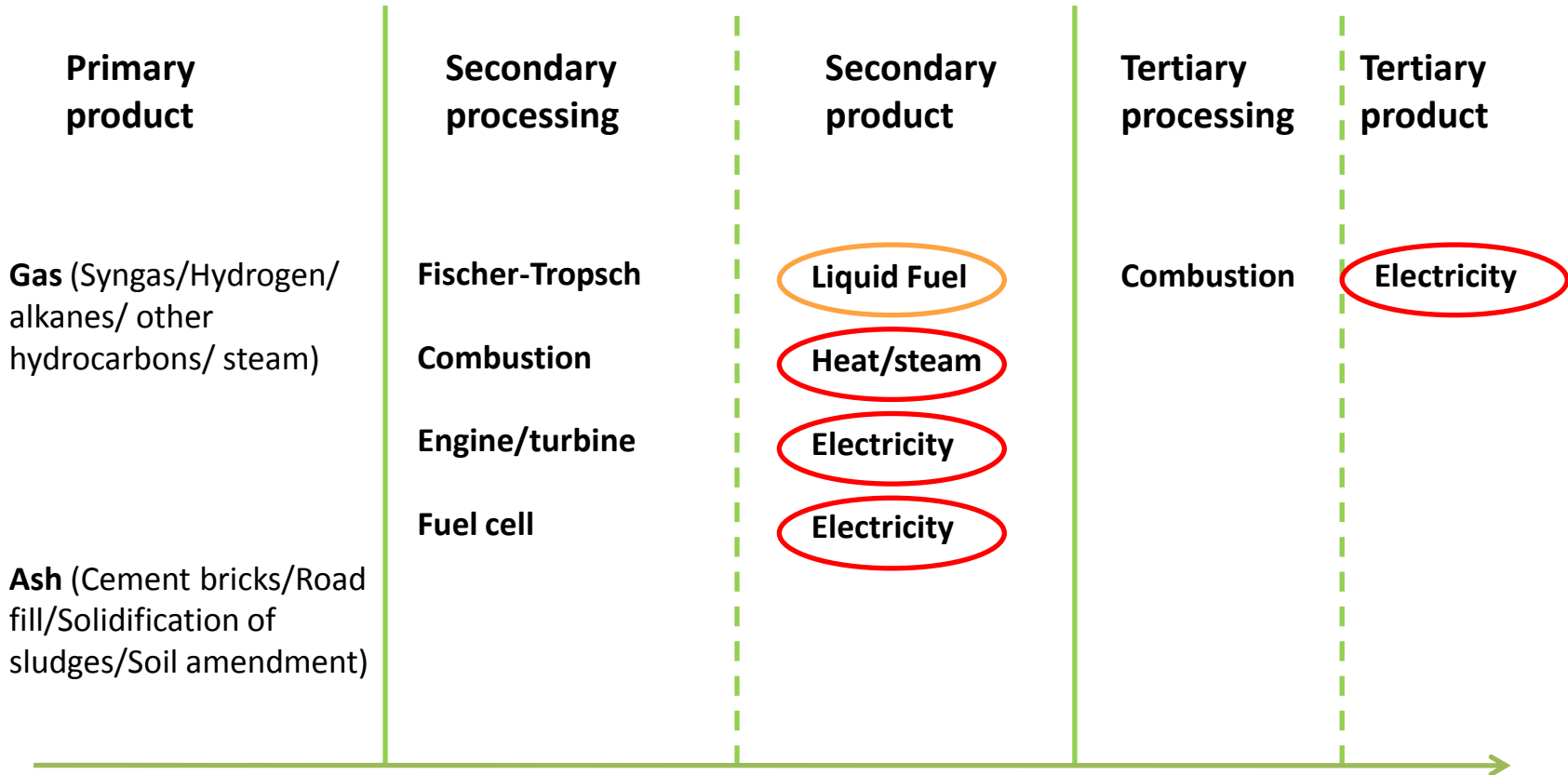


Applications of thermo-chemical processes

Primary product	Secondary processing	Secondary product
Bio-oil	Combustion	Heat/Steam
	Engine/Turbine	Electricity
	Stabilization	Stabilized oil
	Upgrading	Hydrocarbons Chemicals (Fertilizer/Surfactant/ Plasticizers/Octane enhancer/Chemical precursors/Polymers)
	Extraction	Chemicals
	Gasification	Gas
Stage 1	Stage 2	Stage 3



Applications of thermo-chemical processes



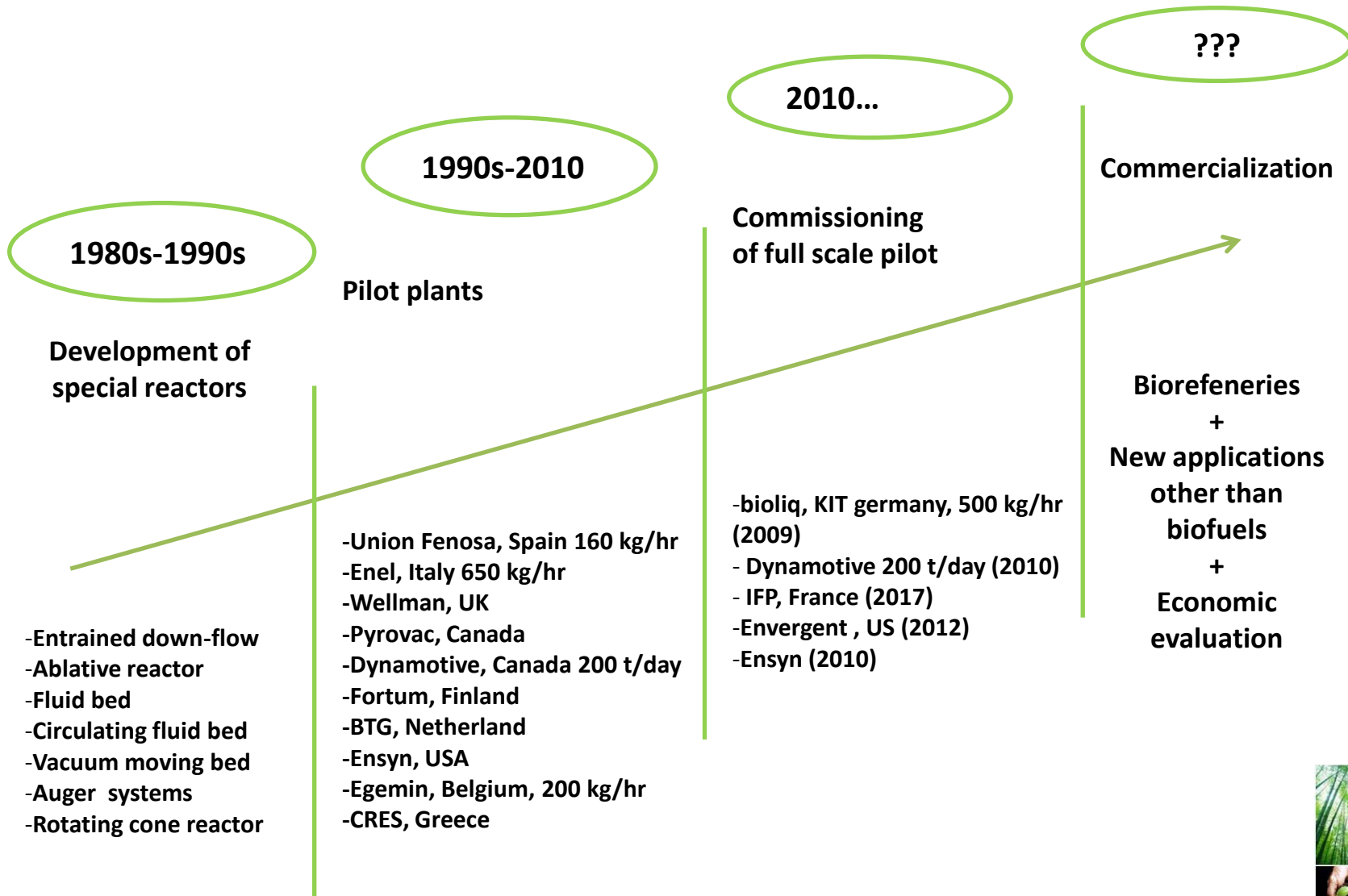
Stage 1

Stage 2

Stage 3



Commercial plants

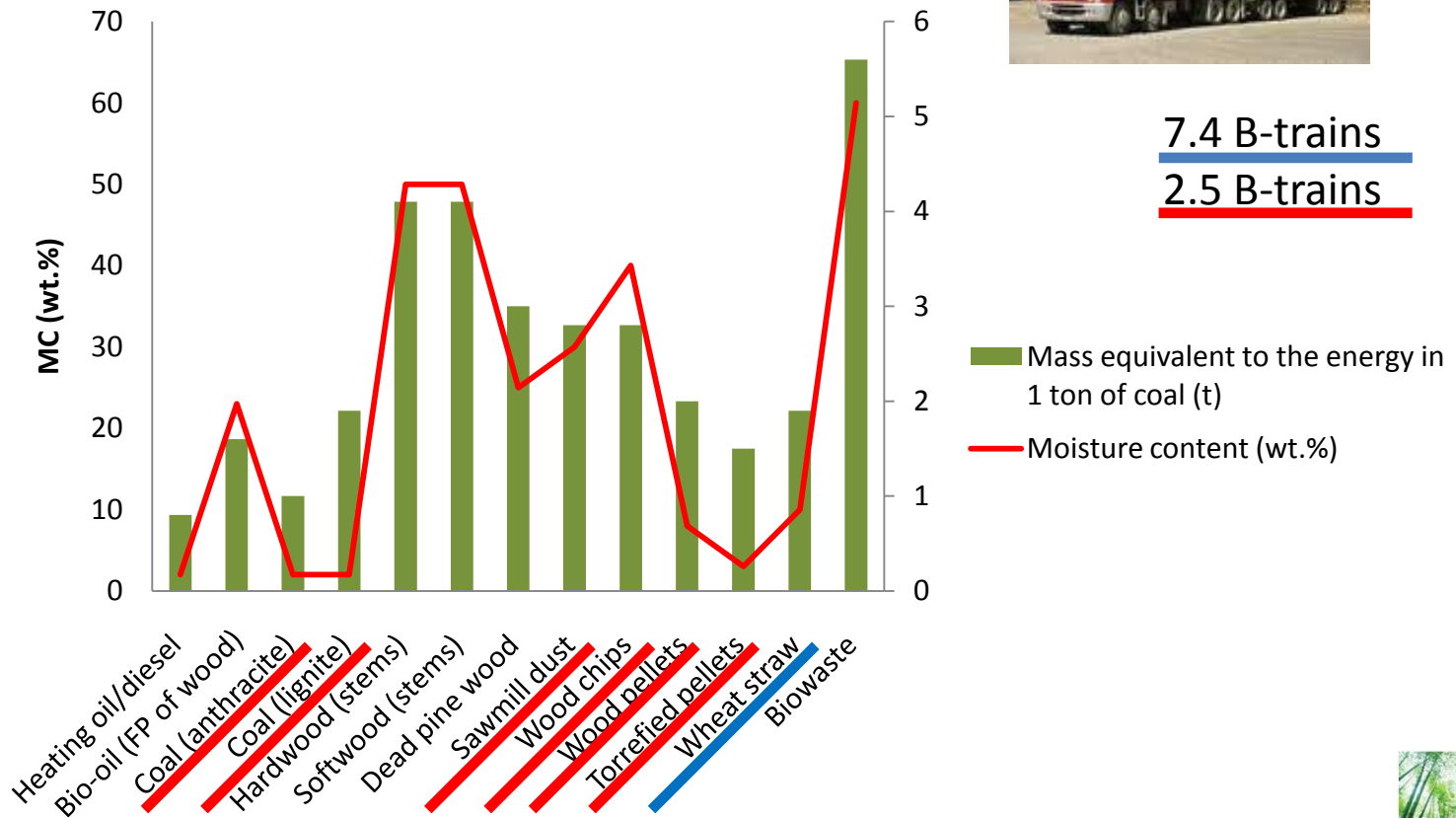


Keys-Challenges

- **Thermo-chemical processes more adapted** than bio-conversion to convert **wood** biomasses:
 - Wood contains a high lignin content which is not easily accessible for biological degradation
 - Ethanol production involves multiple steps (pretreatment, enzymatic or acid hydrolysis, fermentation and distillation.
 - 120 L aromatics/m³ = 180 L ethanol/m³
- Large choice of thermo-chemical processes for specific applications:
 - Gasification for synthetic fuel and electricity production
 - Liquefaction for chemicals and fuel production
 - Slow/vacuum for biochar and activated carbon production
 - Fast pyrolysis for fuel production
 - Both, **Chemical refining, derivative products**
- **Transportation:** increase the energy density to decrease the cost of transportation



Transportation



7.4 B-trains

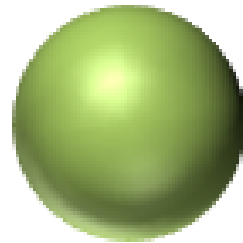
2.5 B-trains



Transportation

Maximum load (t)

Cost (500 km)



Ship

\$0.003/t-km



Train

\$0.03/t-km



Truck

\$0.10/t-km

Densification of wood:

- PELLETIZATION (Small portable pellet unit)
- TORREFACTION (Small portable pyrolysis unit)



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- **Potential of CO₂ sequestration**



CO₂ sequestration



17,475,000 tons in 2008

~ 1,939,090 t CO₂ sequestered

364,157,200 t CO₂ emissions for SA in 2003



Pyrolysis



Bio-oil + Gas



Biochar

Combustion

Soil amendment

Fast pyrolysis and slow pyrolysis, 10 wt.% and 27 wt.% of CO₂ sequestered, respectively

20 wt.% of raw forest wastes - 0.3 wt.% of reduction of CO₂ emissions



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- **Transportation:** increase the energy density to decrease the cost of transportation
- **Potential of CO₂ sequestration**
- Intensification of non-friendly environmental woody culture leading to the **soil impoverishment**



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Thank you for your attention

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